

CHEMISTRY

PHASE TEST –IV
PAPER - 2

Time Allotted: 1 Hr.

Maximum Marks:80

- Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.
- You are not allowed to leave the Examination Hall before the end of the test.

RANKERS STUDY MATERIAL

READ THE INSTRUCTIONS CAREFULLY

QUESTION PAPER FORMAT AND MARKING SCHEME :

1. Section A contains 8 multiple choice questions with one or more than one correct option.
Marking Scheme: +4 for correct answer, 0 if not attempted and –2 in all other cases.
2. Section A contains 2 “Paragraphs” Based on each paragraph, there will be **TWO** questions. Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is(are) correct
Marking scheme: +4 If only the bubble(s) corresponding to all the correct option(s) is(are) darkened, **0** In none of the bubbles is darkened, **–2** In all other cases
3. Section **C** contains 8 questions. The answer to each question is a single digit integer ranging from 0 to 9 (both inclusive).
Marking Scheme: +4 for correct answer and 0 in all other cases.

Name of the Candidate

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Enrolment No.

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Useful Data

Gas Constant	R	=	8.314 J K ⁻¹ mol ⁻¹
		=	0.0821 Lit atm K ⁻¹ mol ⁻¹
		=	1.987 ≈ 2 Cal K ⁻¹ mol ⁻¹
Avogadro's Number	N_a	=	6.023 × 10 ²³
Planck's constant	h	=	6.625 × 10 ⁻³⁴ J·s
		=	6.625 × 10 ⁻²⁷ erg·s
1 Faraday		=	96500 coulomb
1 calorie		=	4.2 joule
1 amu		=	1.66 × 10 ⁻²⁷ kg
1 eV		=	1.6 × 10 ⁻¹⁹ J

Atomic No: H=1, He = 2, Li=3, Be=4, B=5, C=6, N=7, O=8, N=9, Na=11, Mg=12, Si=14, Al=13, P=15, S=16, Cl=17, Ar=18, K =19, Ca=20, Cr=24, Mn=25, Fe=26, Co=27, Ni=28, Cu = 29, Zn=30, As=33, Br=35, Ag=47, Sn=50, I=53, Xe=54, Ba=56, Pb=82, U=92.

Atomic masses: H=1, He=4, Li=7, Be=9, B=11, C=12, N=14, O=16, F=19, Na=23, Mg=24, Al = 27, Si=28, P=31, S=32, Cl=35.5, K=39, Ca=40, Cr=52, Mn=55, Fe=56, Co=59, Ni=58.7, Cu=63.5, Zn=65.4, As=75, Br=80, Ag=108, Sn=118.7, I=127, Xe=131, Ba=137, Pb=207, U=238.