Chemistry

Straight Objective Type

This section contains 30 multiple choice questions numbered 1 to 30. Each question has 4 choices (A), (B), (C) and (D), out of which **ONLY ONE** is correct.

1. What is the correct decreasing order of enol content of the following compounds

CH₃CHO

- (B) $CH_3CHO > CH_3COCH_2CO_2Et > CH_3COCH_2CHO > PhCOCH_2COPh$
- (C) CH₃CHO > CH₃COCH₂CHO > PhCOCH₂COPh > CH₃COCH₂CO₂Et
- (D) None
- 2. What is the correct increasing order of acidicity of the following compounds?
 - (A) $CH_3NH_2 > C_5H_5N > CH_3CN$
- (B) $CH_3NH_2 < C_5H_5N < CH_3CN$
- (C) $CH_3NH_2 < CH_3CN < C_5H_5N$
- (D) $CH_3CN < CH_3NH_2 < C_5H_5N$

- 3. Allyl isocyanide has
 - (A) 9 σ bonds and 4 π bonds
 - (B) 9 σ bonds, 3 π bonds and 2 non bonded electrons
 - (C) 8 σ bonds and 5 π bonds
 - (D) 8 σ bonds, 3 π bonds and 4 non bonded electrons.

4. CHO
(A) + 6 membered ring (B) . B is

(A)
$$(CI)$$
(C) (CI)
(C

- 5. Which of the following molecule having maximum dipole moment?
 - (A) (B) (D) (D)

6. OH
$$(A) O (B) O$$

$$(C) (D) O Me$$

Rough Work

7. Which of the following will undergo haloform reaction?

8.

HO-N N-OH
$$0 \ge N$$
 N=O $N = 0$ (II)

The correct statement about the compound(s) (I) and (II) are:

- (1) I shows geometrical isomerism
- (2) II shows geometrical isomerism
- (3) I and II are functional isomers
- (4) I can show tautomerism
- (A) 1 and 2

(B) 1, 2 and 3

(C) 2 and 4

- (D) 1, 2, 3 and 4
- 9. How many stereoisomers are possible for the given compound

- (A) 2
- (C) 6

- (B) 4
- (D) 8
- 10. Which of the following order are correct?
 - I. Acidity order: o-nitrobenzoic acid > p-nitrobenzoic acid > m-nitrobenzoic acid
 - II. Basicity order: $NH_2^- > EtO^- > OH^- > RCOO^- > CI^-$
 - III. Heat of hydrogenation: cis-2-butene > trans-2-butene
 - IV. Ease of decarboxylation :

$$C_6H_5COCOOH>Ph-C-CH_2-COOH>Ph-CH-COOH>Ph-CH-COOH$$

(A) I and II

(B) I and III

(C) I and IV

(D) I, II and III

$$\begin{array}{c} \text{NH}_2 \\ \text{H}_2\text{C} \\ \end{array} + \begin{array}{c} \text{NH}_2 \\ \text{CH}_3 \\ \end{array}$$

(A)

(C)
$$\begin{array}{c} \text{NH}_2 & \text{O} & \text{O} \\ \text{C} - \text{CH}_2 - \text{C} - \text{CH}_3 \end{array}$$

(B)

(D)

12.

$$\begin{array}{c}
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\
 & & \\$$

A is:

(C) Br (B)

13.

Which of the following set of statements is correct?

- (i) The reactivity order of C H bond is Y > Z > X > W
- (ii) The max. no. of monochloro isomers possible is 4
- (iii) Monochloro isomers on fractional distillation will separate in 4 coloums
- (A) Only (ii) correct

(B) (i) and (ii) correct

(C) (i) and (iii) correct

(D) All are correct

(B) Ph—C——C——OMe

OMe

Which of the following set is correct for the above reaction

- (I) the above reaction will yield the alkene as one of the product
- (II) the above reaction will give both cis and trans isomers
- (III) the above reaction will give cis isomer only
- (IV) the above reaction will give trans isomer only
- (A) (I), (II)

(B) (I), (III)

(C) (I), (IV)

(D) Only (II)

16. Which of the following sequence of reagents can be best utilized for the given transformation

17. Which of the following is the most likely product from the reaction illustrated by the curved arrows in the formula?

Rough Work

:ÓH

- 19. Which of the following is the most stable resonance structure?
- 20. A chiral $C_5H_{10}O$ ether reacts with hot HI to give a $C_5H_{10}I_2$ product. Treatment of this with hot KOH in ethanol produces 1,3-pentadiene. What is the structure of the original ether?



(B)

(D)

21. Which of the following would work best for the synthesis of the ether shown below?

$$O_2N$$
 O_2N
 O_2N
 O_2N

 O_2N F + ONa NO_2

(C) O_2N NaO + F NO_2

- O₂N + HO — NO₂ → NO₂
- 22. In a Cannizaro reaction, the intermediate that will be the best hydride donor is:

(A) OH O⊖

(B) 0[©]

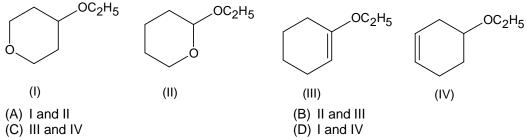
(C) H₃CO

(D) O₂N

23. Ninhydrin reagent reacts with α -amino acids to give purple colour. In the reaction of ninhydrin with phenylalanine, which of the following is responsible for this colour?

(A)
$$H_5C_6-CH_2-CH=CH-C_6H_5$$
 (B) $C_5-C_6H_5$ (C) C_6H_5 (D) $C_6-C_6H_5$

24. Formulas for four ethyl ethers are drawn below. Two are cleared by aqueous acid much more easily than the other two. Which ethers are more easily hydrolysed?



25. $CH_3 \xrightarrow{Zn(Hg)} (A)$; Identify A? CH_2-CH_3 (B) CI

Rough Work

- 26. Glycerol on oxidation with Fenton's reagent (H₂O + FeSO₄) gives:
 - (A) Glyceraldehyde

(B) Dihydroxy acetone

(C) Tartaric acid

- (D) Glyceraldehyde and dihydroxy acetone
- 27. Aniline when treated with NaNO₂ and HCl at 0°C gives
 - (A) phenol

(B) nitrobenzene

(C) benzene diazonium chloride

(D) benzyl alcohol

28. An ether (A) C₆H₁₄O when heated with excess of hot conc. HI produced two alkyl halides which on hydrolysis form compounds (B) and (C). Oxidation of (B) gave on acid (D) where as oxidation of (C) gave a ketone (E). What are structural formula of (A), (B) and (C) respectively?

(A)
$$H_5C_2$$
—O—CH—CH $_2$ -CH $_3$, CH_3 CH $_2$ OH

(B) $C_2H_5OCH_2C_3H_7$, C_2H_5OH , OHC – $CH_2CH_2CH_3$

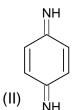
(C)
$$H_3$$
C- CH_2 - CH_2 - CH_2 - CH_3 $HO-CH-CH_3$, CH_3 CH $_2$ CH $_2$ OH CH_3

(D)
$${\rm H_3C-CH-O-CH_2-CH-CH_3}, {\rm H_3C-CH-OH}, {\rm CH_3CH_2CH_2OH} \\ {\rm CH_3} {\rm CH_3} {\rm CH_3}$$

ΝH

29. Tautomerism is exhibited by

- (A) (I) and (III)
- (C) (I), (III) and (IV)



- (III)
- (B) (I), (II) and (IV)
- (D) (IV) only
- 30. Identify the major product in the following reaction

(A)

(C) A and B in equal concentrations

- (B)
- (D) CH_3